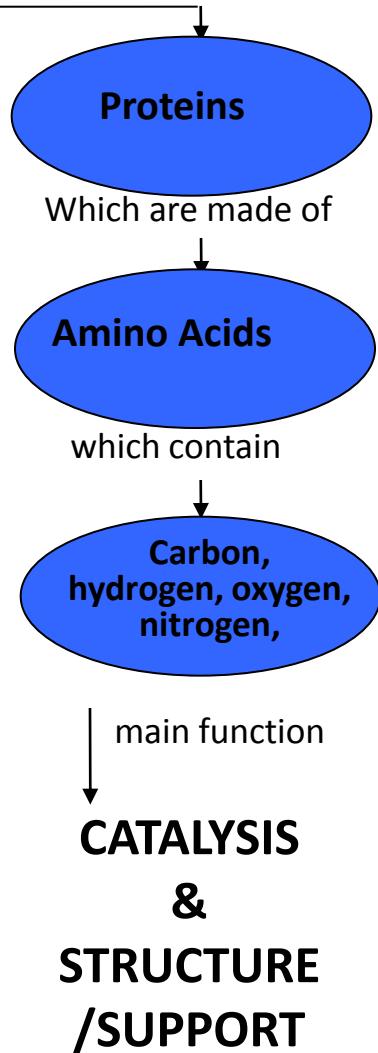
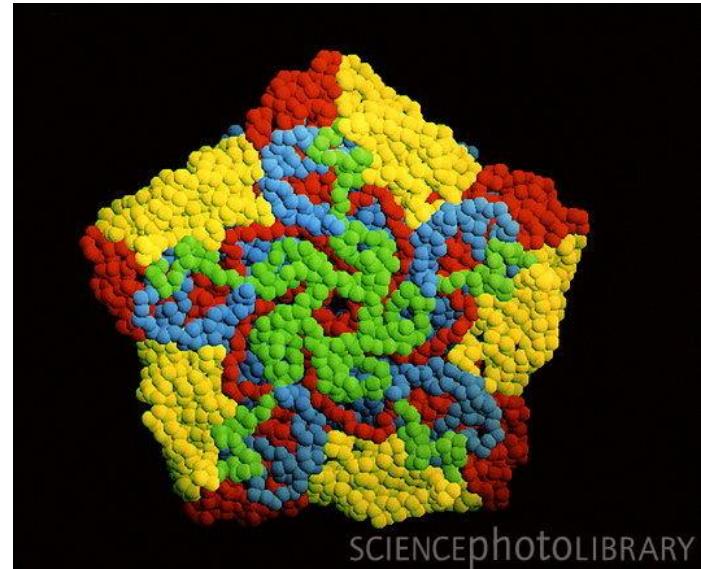
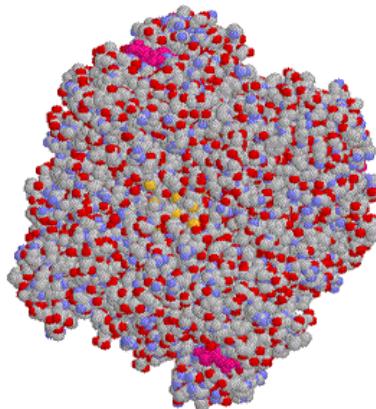


Proteins

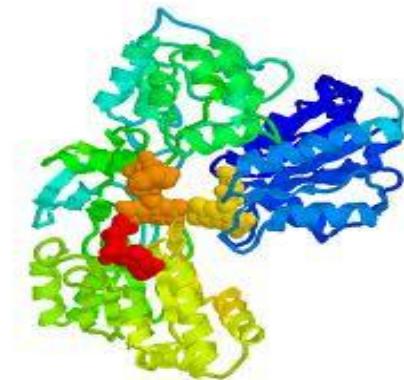
Proteins are essential parts of living organisms and participate in virtually every process in cells.



This is the Enzyme Catalase



Protein coat of polio virus



Types

Function/Example

Enzymatic

Acceleration of chemical reactions

Catalytic

E.g., digestive enzymes (lactase, amylase), cellular respiration

Structural

Collagen & elastin, keratin in hair and nails

Transport

Transport of other substances

E.g., hemoglobin transports O₂ to cells

Hormonal

Cellular communication

E.g., insulin secreted by the pancreas

Contractile

Movement

E.g. actin and myosin in muscle cells

Defensive

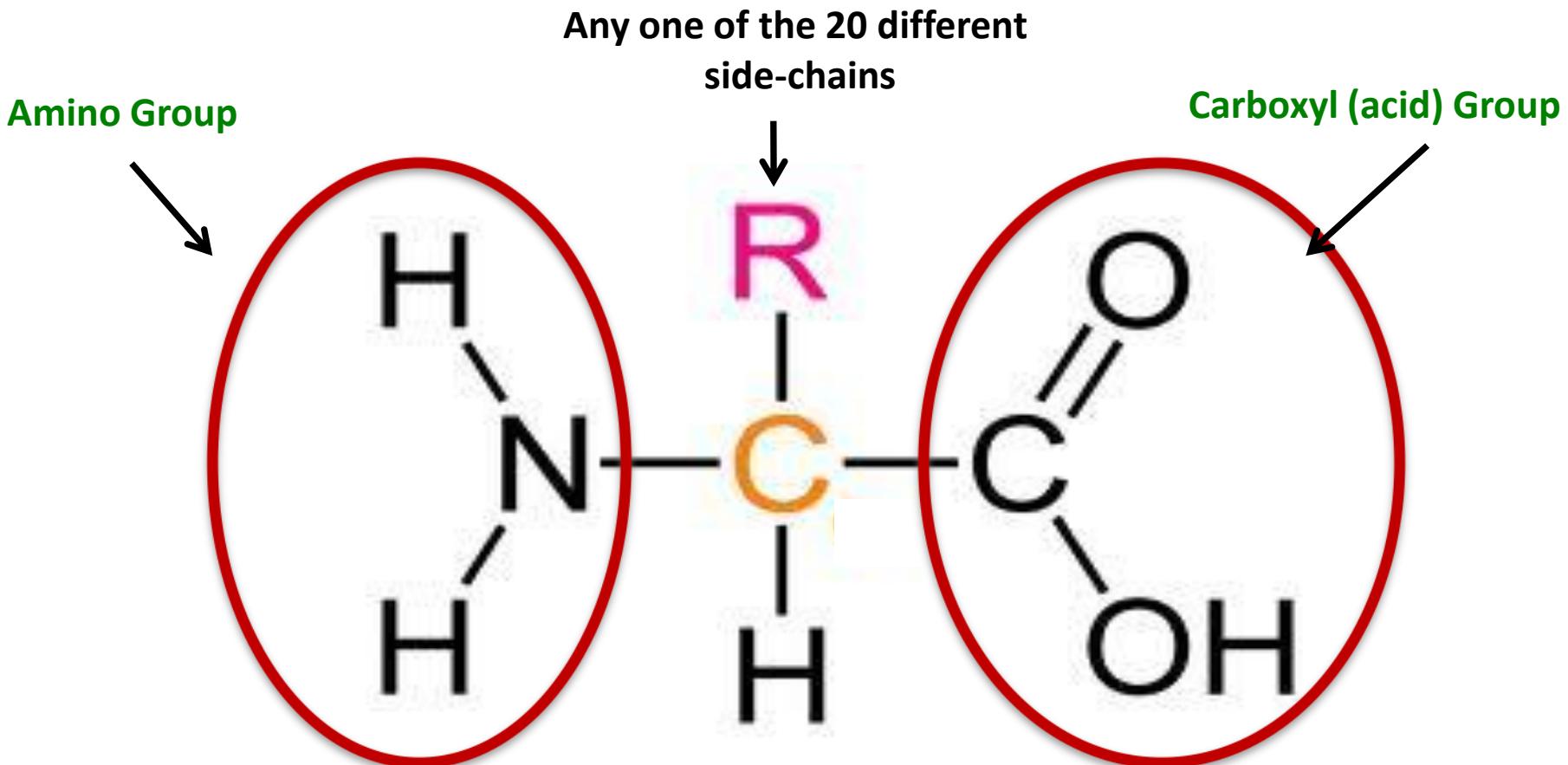
Protect against disease

E.g., antibodies (such as immunoglobulin) combat viruses and bacteria

Proteins and their subunits

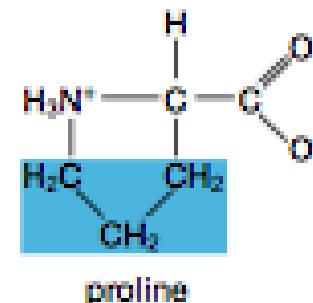
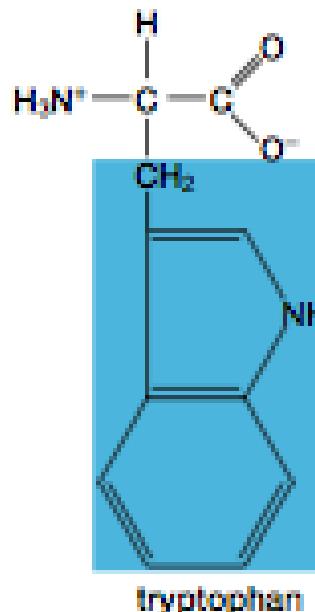
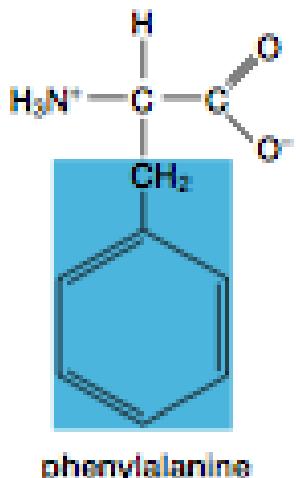
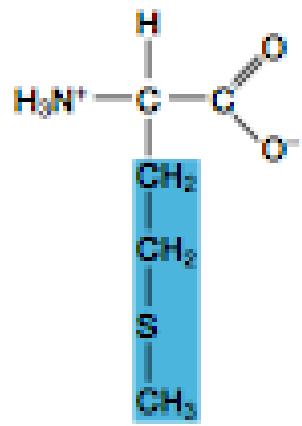
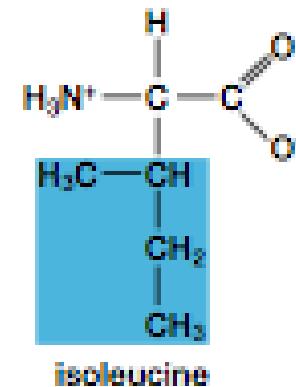
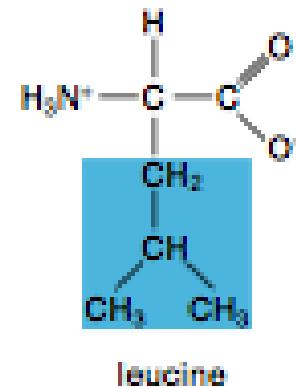
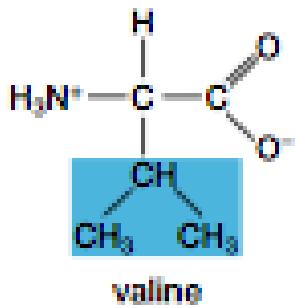
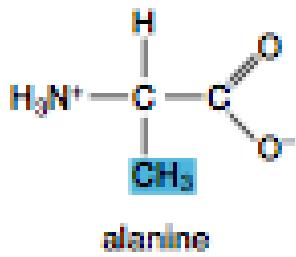
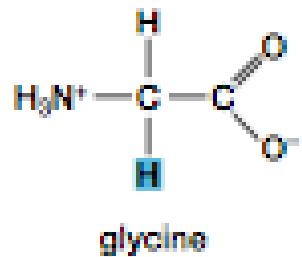
Amino acids are the building blocks of proteins

Amino Acid Structure



Proteins and their subunits

Examples of amino acids



Proteins and their subunits

20 Major Amino Acids

8 are considered “essential”

1. Phenylalanine
2. Valine
3. Threonine
4. Tryptophan
5. Isoleucine
6. Methionine
7. Leucine
8. Lysine

The other 12

1. Glycine
2. Alanine
3. Proline
4. Serine
5. Cysteine
6. Asparagine
7. Glutamine
8. Histidine
9. Tyrosine
10. Aspartic acid
11. Glutamic acid
12. Arginine

Types of Amino Acids

Nonpolar

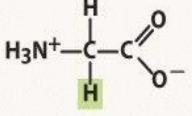
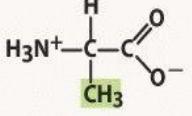
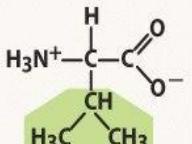
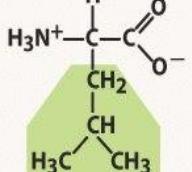
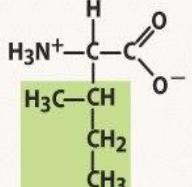
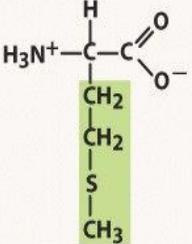
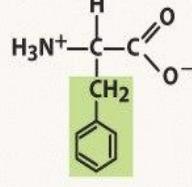
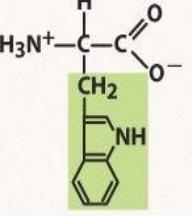
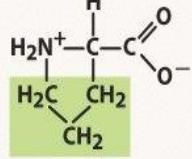
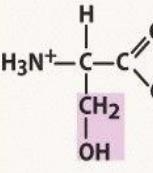
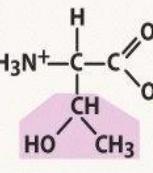
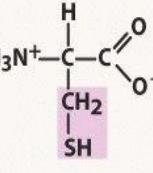
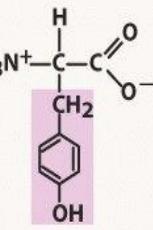
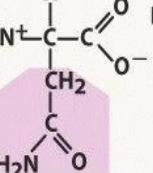
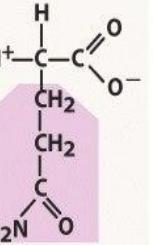
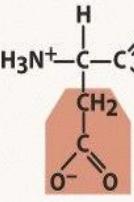
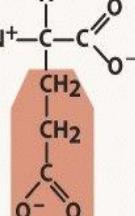
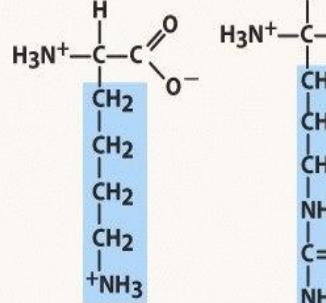
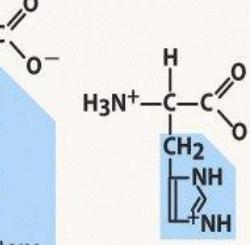
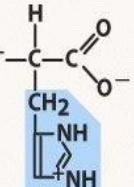
Polar

Polar/Acidic

Polar/Basic

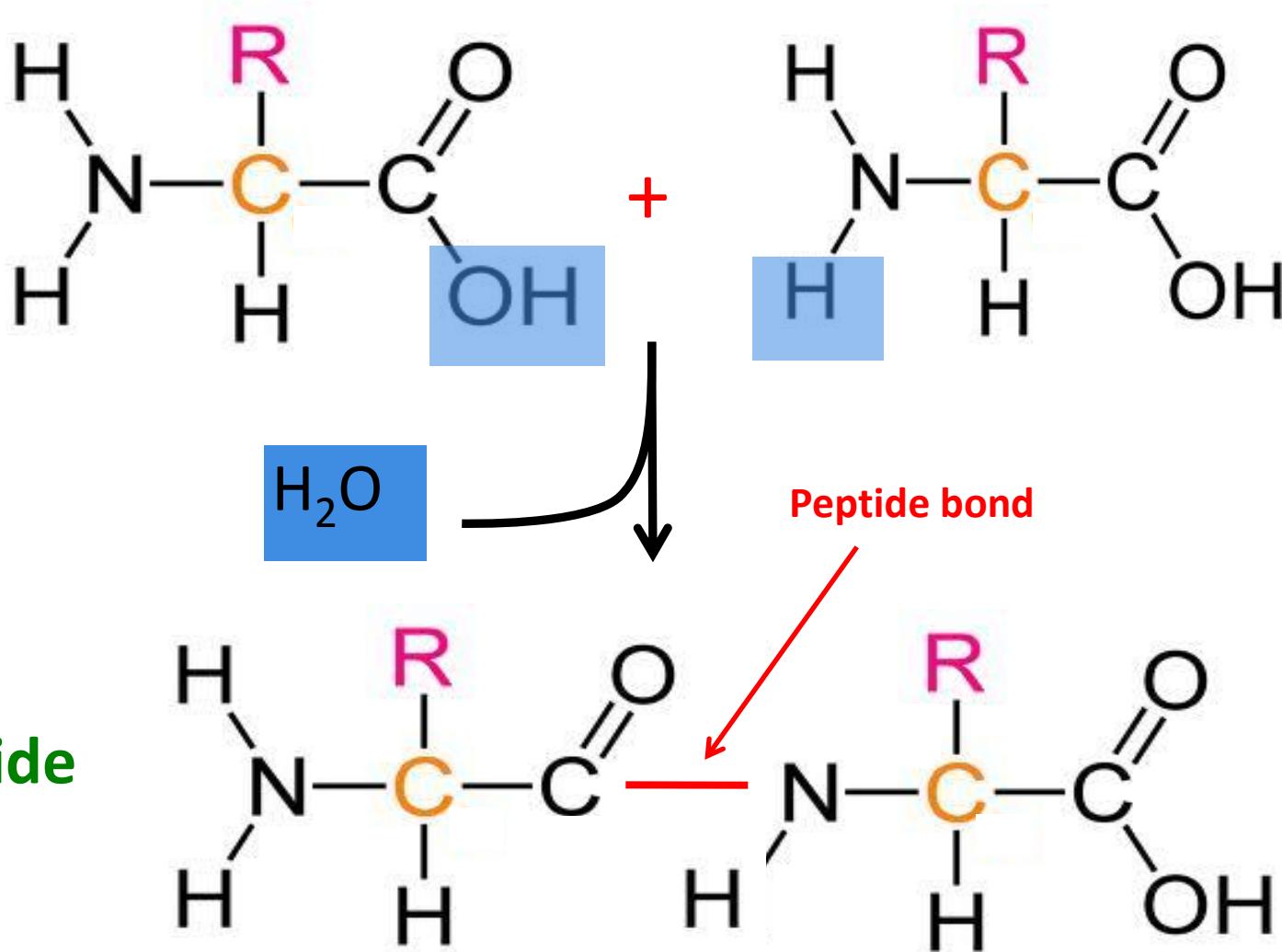
Amino acids each have their own unique chemical properties.

Some dissolve in water – some do not.
This is essential for transport and storage.

					
Nonpolar side chains	Glycine (G) Gly	Alanine (A) Ala	Valine (V) Val	Leucine (L) Leu	Isoleucine (I) Ile
					
	Methionine (M) Met	Phenylalanine (F) Phe	Tryptophan (W) Trp	Proline (P) Pro	
Polar side chains					
	Serine (S) Ser	Threonine (T) Thr	Cysteine (C) Cys	Tyrosine (Y) Tyr	Asparagine (N) Asn
					
					Glutamine (Q) Gln
	Acidic		Basic		
Electrically charged side chains					
	Aspartate (D) Asp	Glutamate (E) Glu	Lysine (K) Lys	Arginine (R) Arg	Histidine (H) His

Making and Breaking Proteins

Amino acids are linked together by peptide bonds
- a special covalent bond found in proteins



Making and Breaking Proteins

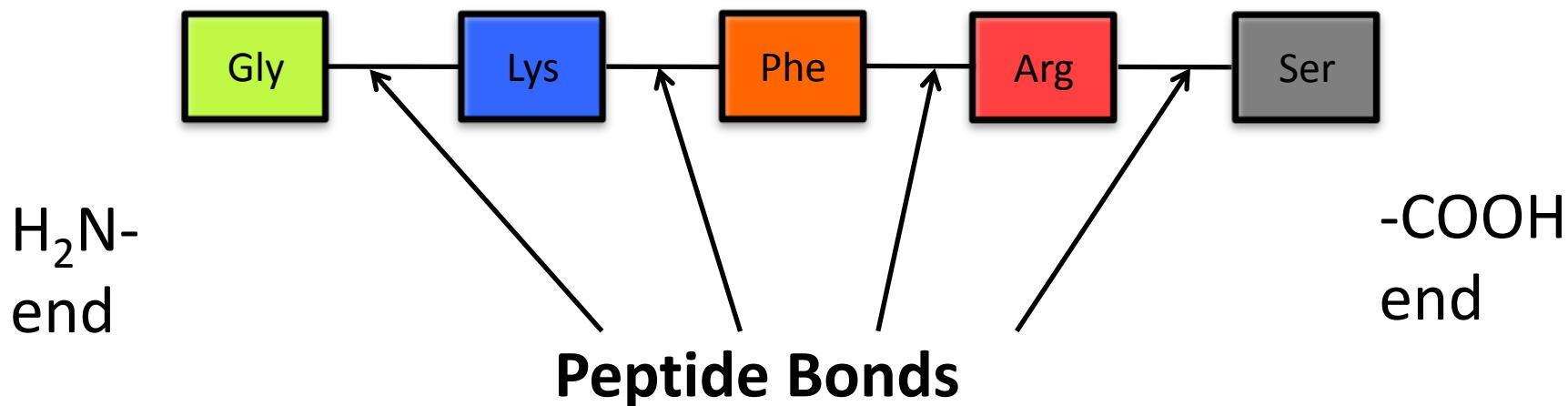
Condensation synthesis

- two amino acids join (dipeptide)
- a peptide bond is formed
- a water molecule is formed

Hydrolysis

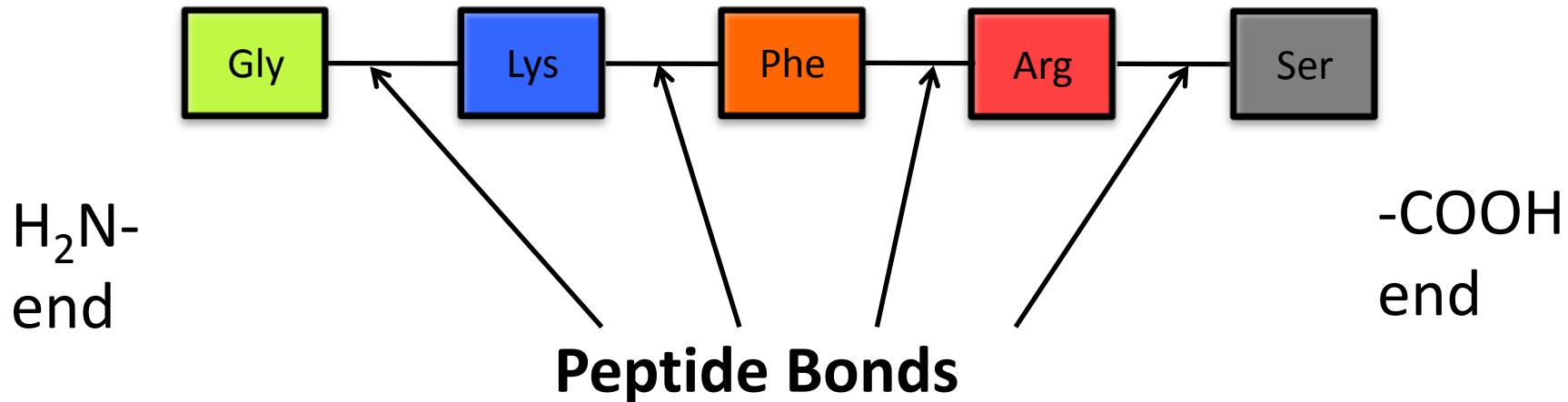
- water is added
- a peptide bond is broken
- amino acids are split apart

A chain of amino acids is called a **polypeptide**



Making and Breaking Proteins

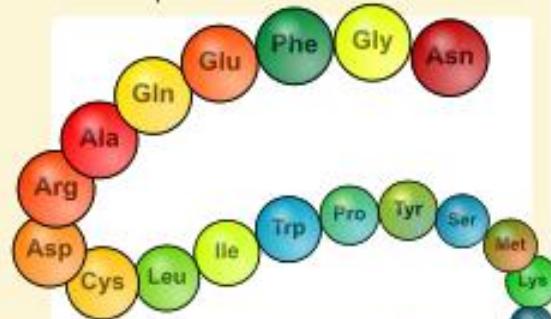
A chain of amino acids is called a **polypeptide**



The type of protein is determined by:

- ✓ sequence of polypeptides
- ✓ orientation in space
- ✓ 3-D shape

Primary structure
amino acid sequence



Four levels of protein structure:

Primary - exact sequence of amino acids before folding.

Secondary - simple folding creates simple structures.

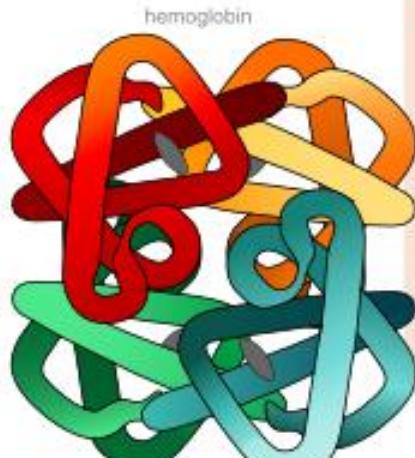
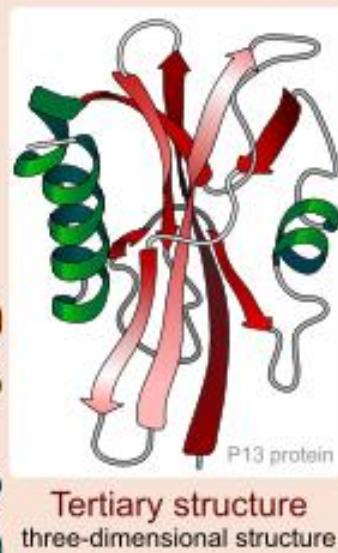
Tertiary - folding results in complex 3D structures.

Quaternary - multiple 3D subunits organized into a bigger structure.

Sulfhydryl (-SH) functional groups

can form disulfide (-S-S) bonds which contribute to a protein's tertiary structure.

Secondary structure
regular sub-structures

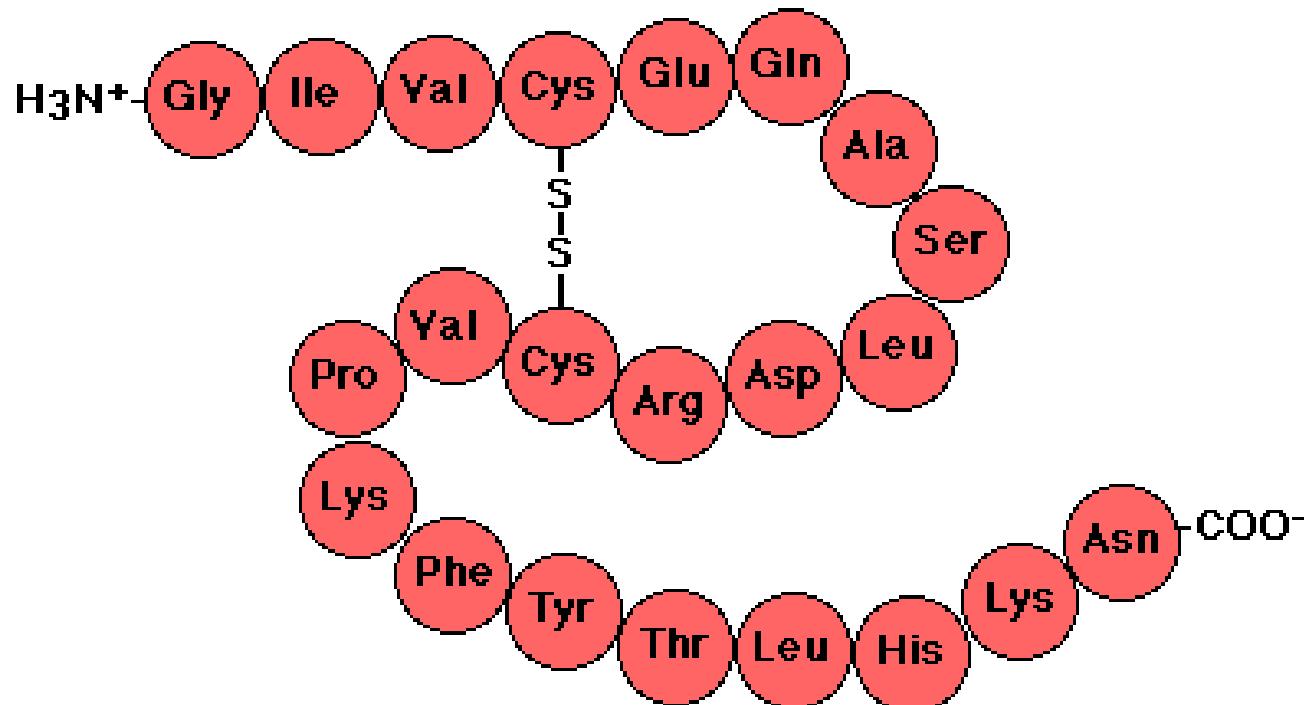


Quaternary structure
complex of protein molecules

7.5 Proteins – Primary Structure (1°)

The amino acid sequence

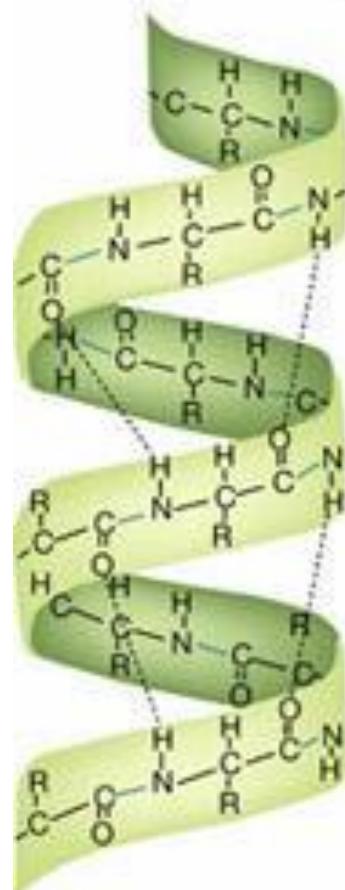
Proteins differ in the variety, number and order of amino acids



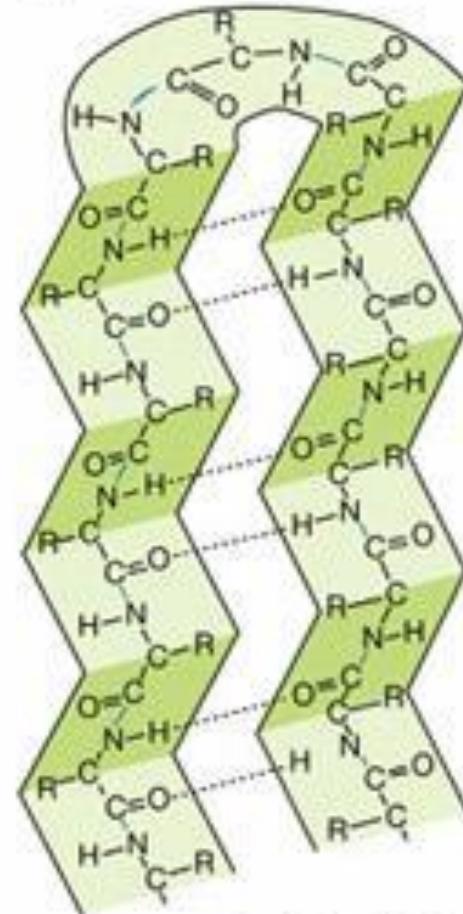
7.5 Proteins – Secondary Structure (2°)

- The way the polypeptide chain is shaped
- Due to formation of H-bonds between the carboxyl O of one aa and the amino H of another aa
- Does not involve R-groups

Secondary structure is the result of hydrogen bonding



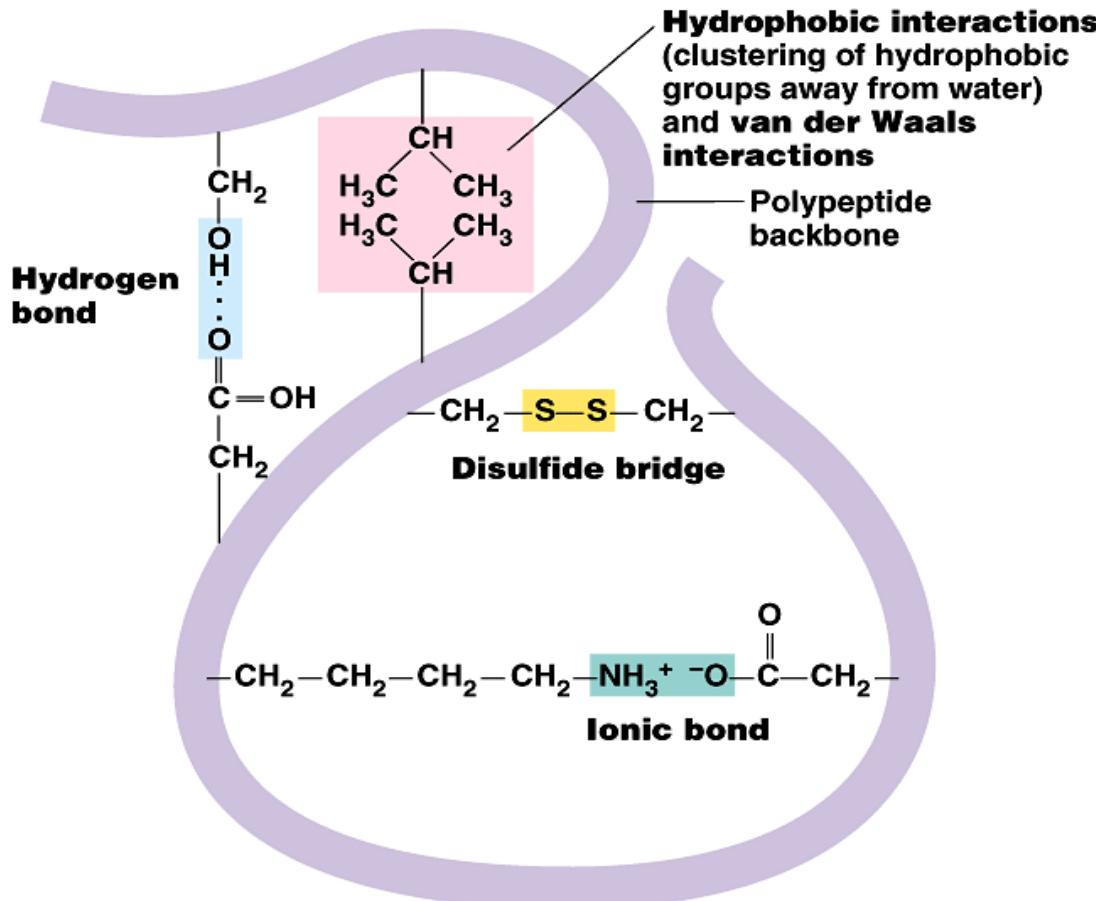
α -helix



β -pleated sheet

7.5 Proteins – Tertiary Structure (3°)

- Compact folding due to interactions between R-groups
- Unique to each proteins
- Important for determining specificity of enzymes



4 bonding types:

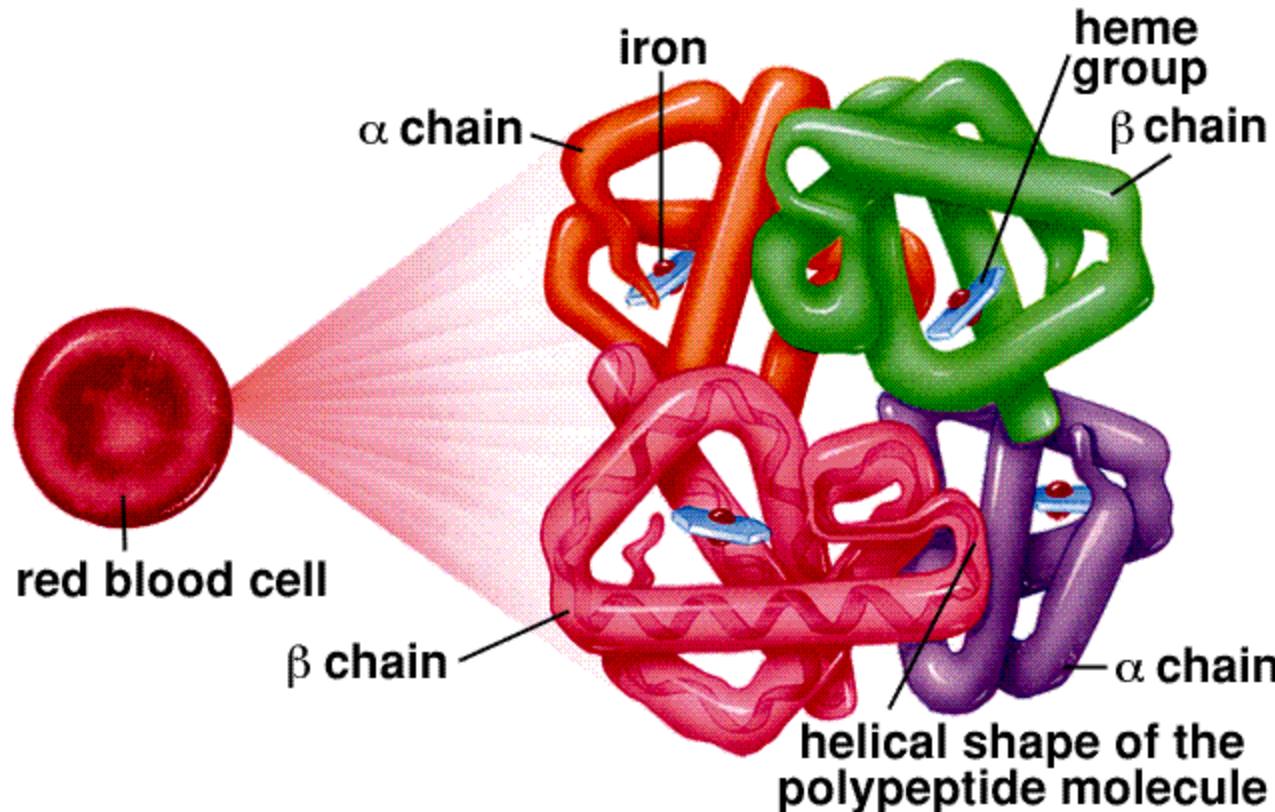
- 1) H-bonds between polar side chains
- 2) Van derWaals forces among hydrophobic side chains
- 3) Covalent bonds between two -SH groups of cysteine side chains → forms disulfide bridges
- 4) Ionic bonds between + and - side chains

7.5 Proteins – Quaternary Structure (4^o)

- Two or more polypeptide chains together
- Often has non-polypeptide groups → conjugated protein
- Not all proteins have a quaternary structure

Sylvia S. Mader, Inquiry into Life, 8th edition. Copyright © 1997 The McGraw-Hill Companies, Inc. All rights reserved.

Hemoglobin Molecule



Hemoglobin

Carries oxygen in the blood

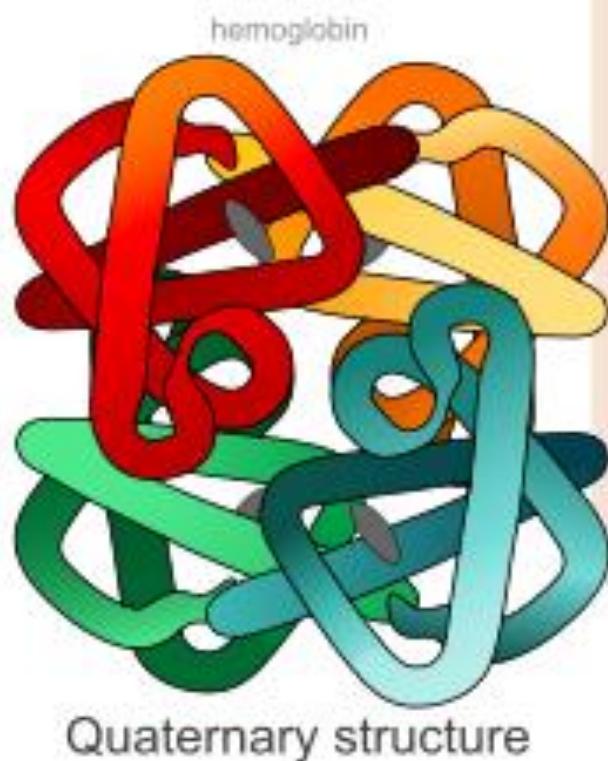
- It's made up of 4 specific 3D subunits

**Proper protein function depends
on correct 3D structure.**

**Any change in the specific primary structure
can cause the protein to fold differently.**

**A different shape can lead to a
different function
(or lack of proper function).**

Sickle cell anemia is an example.

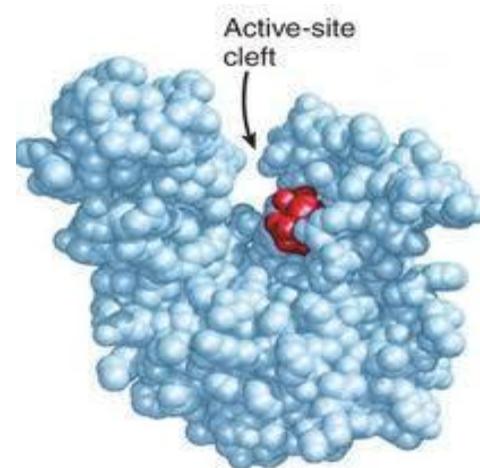
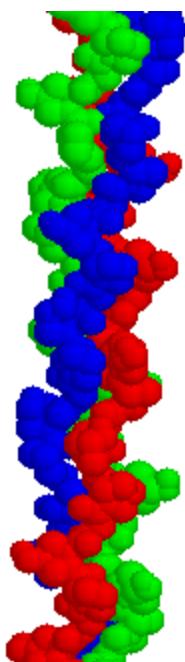


7.5 Protein Types

Globular Proteins

Water soluble and have binding sites and tend to move around.

Examples include hemoglobin and insulin



Fibrous Proteins

Insoluble in water and are involved in structure or movement.

Examples include collagen, actin, and microtubules

7.5 Polar and Non-polar Amino Acids

➤ AA's with **polar** R-groups = hydrophilic

➤ Linked to the hydrophilic areas of the cell membrane

➤ Create hydrophilic channels through the cell membrane which allow polar substances to move through cell membranes.

➤ Found on outside of cell membrane

➤ AA's with **non-polar** R-groups =

hydrophobic

➤ Linked to the hydrophobic areas of the cell membrane

➤ Polar and non-polar aa's are important in determining the specificity of an enzyme. They act as enzyme binding sites (active sites)

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